

# Distribution of actinomycetes in different soil ecosystems and effect of media composition on extracellular phosphatase activity

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## Abstract

Actinomycetes are widely distributed in different habitats and involved in important processes. Therefore, evaluation of their distribution is important in understanding their ecological role. Ninety seven samples were taken from different soil ecosystems (forest, pasture, rain-fed and irrigated cultivated land) located in various climatic zones in the province of Golestan, Northeast Iran. The number of actinomycetes as well as pH value, organic carbon (OC) and soil salinity expressed as electrical conductivity (EC) showed significant differences in the soil ecosystems under investigation. The number of actinomycetes decreased from  $2.86 \times 10^6$  cfu g<sup>-1</sup> in irrigated cultivated land to  $7 \times 10^5$  cfu g<sup>-1</sup> in pasture ( $p < 0.01$ ). In pastures, the number of actinomycetes was negatively correlated with EC ( $r = -0.60$ ,  $n = 15$ ,  $p < 0.001$ ) and pH ( $r = -0.59$ ,  $n = 15$ ,  $p < 0.001$ ). In other soil ecosystems under investigation no significant correlations between soil pH, OC and EC and the number of actinomycetes could be found. In forest soils, the number of actinomycetes was significantly dependent ( $p < 0.05$ ) on climate. A significantly higher population was observed under semi-arid conditions compared to all other climatic zones (humid, sub-humid, Mediterranean). Because many soil microorganisms are able to transform different forms of insoluble organic and inorganic phosphorus into a soluble form suitable for plant uptake, the capacity of the isolated actinomycetes to secrete phosphatase activity was determined under laboratory conditions. A huge variation in the capacities to produce acid and alkaline phosphatases among the different isolates was observed. These enzyme activities and the capacities to hydrolyze phytate in the fermentation broth were significantly dependent on medium composition. In general, substitution of arginine and glycerol in the modified glycerol arginine medium (MGA) with other nitrogen and carbon sources resulted in a significant reduction of phytate dephosphorylation.

**Keywords:** Actinomycetes, land use, phosphatase, phosphorus, phytate, phytate-degrading enzyme

## 1. Introduction

Phosphorus (P) is an essential element for all organisms and in soil ecosystem phosphorus is often the most limiting nutrient for plant growth. Soil P can exist in organic and inorganic forms. Organic P usually comprises 29-65% of total soil P, but in some organic soils organic P can account for up to 90% of total P (Harrison, 1987). In general the amount of soil P is more than sufficient for the requirement of soil microorganisms and plants. However, only a minor amount of the total soil P is readily available to plants and microorganisms. Therefore, P fertilizers based on rock phosphate have got a very important role in agricultural production. Predictions based on the current use of rock phosphate assume a depletion of global high quality rock phosphate reserves by 2100 (van Vuuren *et al.*, 2010). This concern and the increasing demand for low input agriculture has revealed that a better understanding of the plant-soil-microbial P cycle is of utmost importance.

Numerous reports have been published on the phosphate solubilizing microorganisms (PSM) such as *Bacillus subtilis*, *Pseudomonas* sp. (Rodríguez and Fraga, 1999), *Mesorhizobium* sp. (Peix *et al.*, 2001), *Penicillium* sp. and arbuscular mycorrhizal fungi (Ouahmane *et al.*, 2007). Recently the capability of P solubilization (Ghorbani-Nasrabadi *et al.*, 2012) by actinomycetes was reported. However, most of the studies on actinomycetes have been focused on their capability to solubilize inorganic phosphates. There is only one single study on the capability of actinomycetes to hydrolyze phytate, the predominant form of organic P in soils (Ghorbani-Nasrabadi *et al.*, 2012). Actinomycetes are widely distributed in different habitats and involved in important processes. They are not only able to survive under extreme soil condition such as low level of moisture or high salinity, but actinomycetes are also reported to promote plant growth (Hamdali *et al.*, 2008). Therefore, evaluation of their distribution is important in understanding their ecological role. Several studies have shown that soil properties are more important for bacterial soil

composition than vegetation (Bossio *et al.*, 2005). A comparison among four different land uses (cultivated, pasture, pine plantation and mixed wood forest) in Georgia, USA revealed for example that actinobacteria were more prevalent in pasture and cultivated soils (Lauber *et al.*, 2009). In addition, Burck *et al.* (2003) reported that actinomycetes population is higher in bacterial communities in agricultural land compared to forest soils when analyzing forest, pasture and sugar cane plantations in different countries. Furthermore, actinomycetes population was reported to increase after conversion of land use from forest to agriculture; a behavior common to fungi, but not to bacteria (Burck *et al.*, 2003; Fierer *et al.*, 2009).

The objective of this work was to study the effect of soil properties and land use on actinomycetes population as well as the capability of actinomycetes to produce extracellular phosphate activity (including phytate-degrading activity) and its dependence on medium composition.

## 2. Materials and Methods

### 2.1. Soil sampling

Ninety seven samples were taken from different soil ecosystems (forest, pasture, rain-fed and irrigated cultivated land) located in various climatic zones in the province of Golestan, Northeast Iran.

### 2.2. Measurement of soil chemical and physical properties

Soil samples were air-dried, passed through a 2 mm sieve and thereafter dried in an oven at 105°C for 24 hrs for chemical and physical analysis. Soil texture was determined by the Bouyoucos hydrometer method (Gee and Bauder, 1986). The pH-values of the soils were measured in saturated past (McLean, 1982), whereas electrical conductivity (EC) was determined

in the extract of the water saturated past (Ryan *et al.*, 1999). Organic Carbon (OC) was estimated by the potassium dichromate oxidation method (Nelson and Sommers, 1982).

### 2.3. Enumeration of actinomycetes population

The dilution plate method was used to enumerate common actinomycetes in the soils. Soil samples were air-dried for 72 hrs followed by incubation at 40°C for 16 hours (Williams and Wellington, 1982). The treated soil samples were spread in a serial dilution onto glycerol arginine agar (GAA) (1 g L<sup>-1</sup> L-arginine, 12.5 g L<sup>-1</sup> glycerol, 1 g L<sup>-1</sup> K<sub>2</sub>HPO<sub>4</sub>, 1 g L<sup>-1</sup> NaCl, 0.5 g L<sup>-1</sup> MgSO<sub>4</sub> × 7H<sub>2</sub>O, 0.01 g L<sup>-1</sup> FeSO<sub>4</sub> × 6H<sub>2</sub>O, 0.001 g L<sup>-1</sup> CuSO<sub>4</sub> × 5H<sub>2</sub>O, 0.001 g L<sup>-1</sup> MnSO<sub>4</sub> × H<sub>2</sub>O, 0.001 g L<sup>-1</sup> ZnSO<sub>4</sub> × 7H<sub>2</sub>O, 15 g L<sup>-1</sup> agar (El-Nakeeb and Leachevalier, 1963) supplemented with 100 µg mL<sup>-1</sup> cycloheximide and the Petri dishes were incubated for 14 days at 28°C. Dry powdery or velvety colonies with limiting growth were considered as actinomycetes.

### 2.4. Effect of different nitrogen sources on phytate degradation

Arginine was replaced in the modified glycerol arginine medium (MGA) (1 g l<sup>-1</sup> L-arginine, 12.5 g l<sup>-1</sup> glycerol, 4 g l<sup>-1</sup> Na-phytate, 1 g l<sup>-1</sup> CaCl<sub>2</sub>, 1 g l<sup>-1</sup> NaCl, 0.5 g l<sup>-1</sup> MgSO<sub>4</sub> × 7 H<sub>2</sub>O, 0.01 g l<sup>-1</sup> FeSO<sub>4</sub> × 6 H<sub>2</sub>O, 0.001 g l<sup>-1</sup> CuSO<sub>4</sub> × 5 H<sub>2</sub>O, 0.001 g l<sup>-1</sup> MnSO<sub>4</sub> × H<sub>2</sub>O, 0.001 g l<sup>-1</sup> ZnSO<sub>4</sub> × 7 H<sub>2</sub>O) by different other nitrogen sources ((NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, NaNO<sub>3</sub>, NH<sub>4</sub>NO<sub>3</sub>, malt extract, meat extract, soybean meal) in a concentration of 0.1% (w/v).

Furthermore, glycerol in the MGA medium was replaced by different carbon sources. Simple sugars such as glucose, fructose, sucrose, lactose, and maltose were added in a concentration of 1% (w/v) and complex carbon sources such as pea flour and maltodextrin in a concentration of 0.5% (w/v).

### 2.5. Standard phytate-degrading enzymes assay

Spectrophotometric assay was carried out in a total volume of 2 ml at 37°C for 30 min in 0.1 M Tris-

HCl (pH 7.4) and ammonium acetate (pH 5) buffer containing 10 mM sodium phytate. The enzymatic reactions were started by the addition of 50 µL of enzyme to the assay mixtures. The liberated Pi was measured according to Greiner *et al.* (1998). The reaction was terminated by the addition of 1.5 ml of freshly prepared acetone, 5N sulfuric acid and 10 mM ammonium molybdate reagent (2:1:1 v/v) and there after 100 µl of 1 M of citric acid were added. Absorbance was determined at 355 nm. A calibration curve was prepared from 5 to 600 mM phosphate. Phytate degrading activity (U) was defined as the amount of enzyme that released 1 µmol phosphate per min.

### 2.6. myo-Inositol phosphate analysis by HPLC

Quantification of phytate and other myo-inositol phosphates was performed by ion pair chromatography (Greiner *et al.*, 1998). In brief, the medium was acidified with HCl to give a final concentration of 2.4%. 2 mL of the acidified medium were diluted 1:25 with water and applied to a column (0.7 x 15 cm) containing AG1-X8, 100-200 mesh resin. The column was washed with 25 ml of water and 25 ml of 25 mM HCl. Then myo-inositol phosphates were eluted with 20 ml of 2 M HCl. The eluate obtained was concentrated in a vacuum evaporator to dryness and the residue was solved in 1 mL water. 20 µL of the samples were chromatographed on Ultrasep ES 100 RP18 (2 x 250 mm). The column was run at 45°C and 0.2 mL min<sup>-1</sup> of an eluant consisting of formic acid:methanol:water:tetrabutylammonium hydroxide (44:56:5:1.5 v/v), pH 4.25, as described by Sandberg and Ahderinne (1986). A mixture of the individual myo-inositol phosphate esters (InsP<sub>3</sub>-InsP<sub>6</sub>) was used as a standard.

### 2.7. Acid and alkaline phosphatase assay

Acid and alkaline phosphatase activities were determined according to Greiner *et al.* (2003). 50 µL of each sample was added to 200 µL of 100 mM sodium acetate pH 5 or glycine-NaOH pH 10 containing 1 mM p-nitrophenyl phosphate (pNPP) for acid or

alkaline phosphatase activity, respectively and incubated at 37°C for 30 minutes. The reaction was terminated by addition of 1 ml of 1 M NaOH and thereafter absorbance at 405 nm was determined. Phosphatase activity was defined as the amount of enzyme that released 1  $\mu\text{mol}$  p-nitrophenolate per min.

### 2.8. Effect of media composition on phosphatase activity

To study the effect of the composition of the growth medium on phosphatase activity, five different growth media were investigated. Soya peptone glucose medium (SPG) (25 g L<sup>-1</sup> soybean meal, 60 g L<sup>-1</sup> glucose, 0.143 g L<sup>-1</sup> ZnSO<sub>4</sub>, pH 7.5), yeast extract-malt extract (ISP2) (10 g L<sup>-1</sup> malt extract, 4 g L<sup>-1</sup> yeast extract, 4 g L<sup>-1</sup> glucose, 2 g L<sup>-1</sup> CaCO<sub>3</sub>, pH 7.4), Corn starch medium (20 g L<sup>-1</sup> soybean meal, 40 g L<sup>-1</sup> corn starch, 30 g L<sup>-1</sup> dextrin, 2 g L<sup>-1</sup> (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, 1 g L<sup>-1</sup> (NH<sub>4</sub>)<sub>2</sub>HPO<sub>4</sub>, 8 g L<sup>-1</sup> CaCO<sub>3</sub>, pH 6.8), MGA and Luria-Bertani broth (LB) v (10 g L<sup>-1</sup> tryptone, 5 g L<sup>-1</sup> yeast extract, 5 g L<sup>-1</sup> NaCl, pH 7.0) supplemented with 0.5% rice bran. The isolates were grown in each medium and after 60 hrs phosphatase activity was determined.

### 2.9. Statistical analysis

To elucidate relationships among actinomycetes abundance, land use and climate variation, we compared actinomycete densities from different soil ecosystems. These encompassed four land use, wide range of climates (Arid, semi-Arid, Mediterranean, sub-Humid and Humid), because of differences in elevation and precipitation. Mean annual precipitation and temperature were based on 30 years averages from the meteorological stations and climate classification was based on de Marttone. Geographical information (latitude, longitude and elevation) was based on information from Global Positioning System (GPS). Statistical analysis and mean comparison using Duncan's test were conducted using SAS software (SAS Institute, 2000).

## 3. Results and Discussion

### 3.1. Soil characteristics and actinomycetes population

According to McCarthy and Williams (1992), organic matter, salinity, relative moisture, temperature, pH and vegetation are important factors which control abundance of actinomycetes in soil. A comparison of the mean values of the chemical and physical soil properties as well as the actinomycetes population showed significant differences in pH value, organic carbon (OC), soil salinity expressed as electrical conductivity (EC) and actinomycetes number in the different soil ecosystems studied. Especially, climate variation had a significant effect on all mentioned parameters.

With the exception of forest soils, all studied soil ecosystems had a similar pH value (pH 7.7 - 7.8) (Figure 1a). The pH value of the forest soil was significantly lower (pH 6.92). The pH values found in this study are in good agreement with those reported by Khormali *et al.* (2009). They reported that the pH value increased from 7.2 to 7.5 when land use was changed from forest to agriculture. The apparently lower pH values observed in forest soils are due to a downward movement of basic ions in these soils. Natural processes such as carbon dioxide evolution from plant roots or soil microbial respiration are believed to be responsible for controlling soil pH. Lauber *et al.* (2008) discussed that among others vegetation, soil type and soil management can alter soil pH.

Soil organic matter has a key role on beneficial biological processes as well as on the chemical and physical properties of soils. It provides energy for soil microbial community, increases cation exchange capacity and ameliorates soil aggregate and structure (Wolf and Wagner, 2005). Organic carbon showed significant differences among the studied soil ecosystems. The average OC in forest soils was 4, 5 and 10 times higher than in irrigated, rain-fed and pasture soils, respectively (Figure 1b). These results

Figure 1a

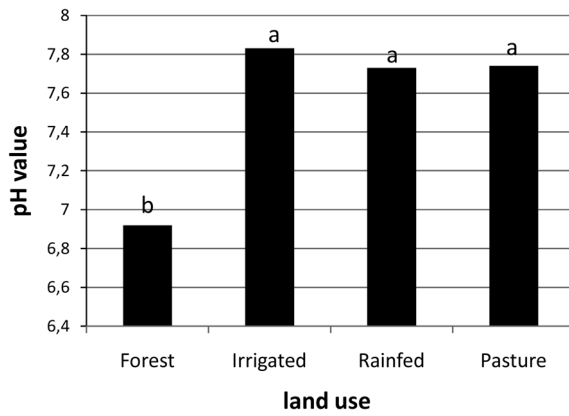


Figure 1b

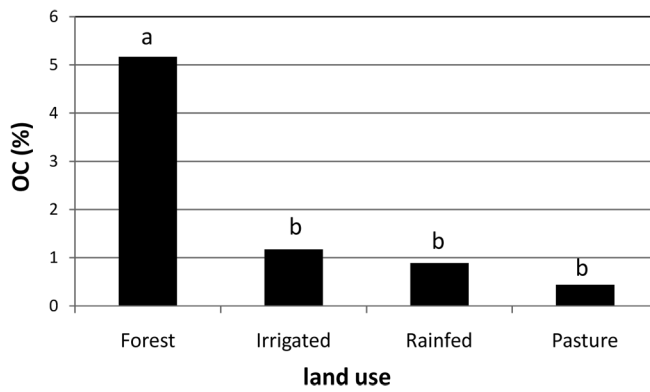
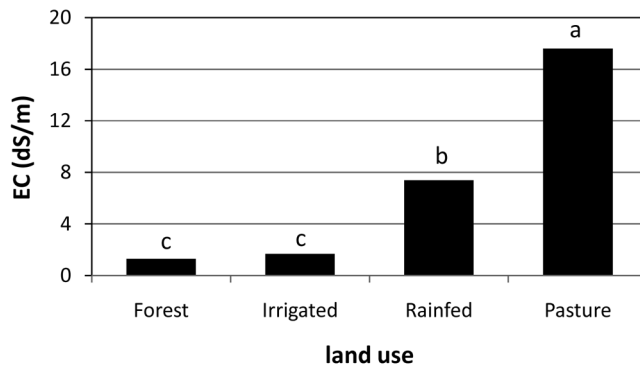
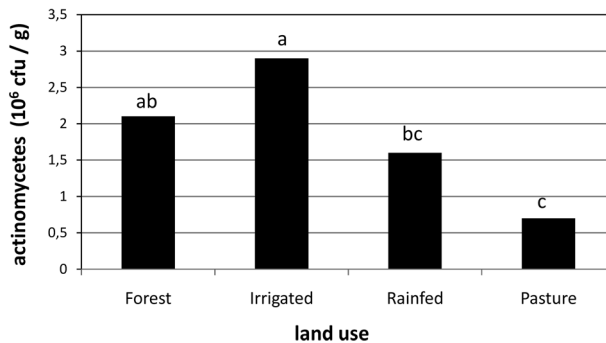


Figure 1c



**Figure 1d**

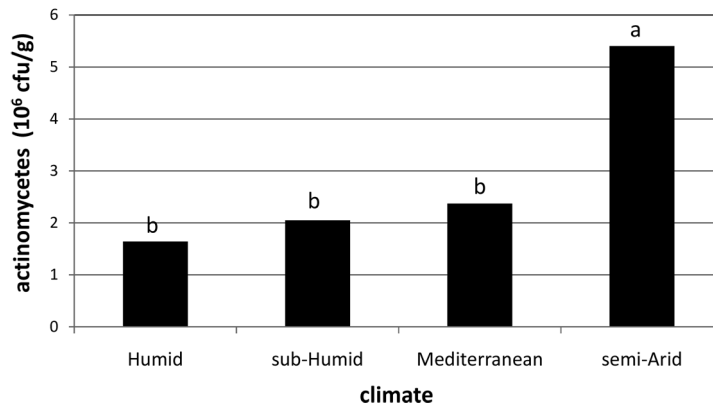
**Figure 1:** Mean values of the chemical and physical soil properties as well as the actinomycetes population in the different soil ecosystems studied (a) pH value, (b) organic carbon (OC), (c) soil salinity expressed as electrical conductivity (EC), (d) actinomycetes abundance columns with identical letters are not statistically different ( $p < 0.05$ )

A strong decline in soil OC occurs after prolonged tillage (Burck *et al.*, 1989), and rupture of the larger aggregates was reported to cause less protection of soil organic matter and to accelerate OC oxidation (Nardi *et al.*, 1996).

Salinity is considered as a natural problem under arid and semi-arid conditions. Sensitive microorganisms are killed by high salt content, whereas some other microorganisms can adapt themselves to saline conditions. In this study, the lowest salinities were found in forest and irrigated soils. Salinity of the rain-fed soil was intermediate and the highest salinity was observed in pasture (Figure 1c). Gennari *et al.* (2007) reported that the number of actinomycetes was  $10^5$  cfu  $g^{-1}$  soil in every soil studied, thus highlighting that the growth of actinomycetes was not affected by the salt content. Our results however suggest that the number of actinomycetes was affected by soil salinity when comparing the irrigated and rain-fed land uses.

To quantify actinomycetes numbers in different soil ecosystems dilution plate counts were used in many studies (Davis and Williams, 1970; Zhang *et al.*, 2000).

Using this methodology, the community size can easily reach more than  $10^6$  cfu  $g^{-1}$  soil which apparently cannot reflect actual numbers of actinomycetes in soil (Goodfellow and Williams, 1983). Nevertheless, the plate counts method was considered to be useful in the evaluation of the abundance of common actinomycetes in soil. Variation of actinomycetes abundance in the different soil ecosystems studied is shown in Figure 1d. The number of actinomycetes decreased from  $2.86 \times 10^6$  cfu  $g^{-1}$  in irrigated cultivated land to  $7 \times 10^5$  cfu  $g^{-1}$  in pasture ( $p < 0.01$ ). In pastures, the number of actinomycetes was negatively correlated with soil salinity ( $r = -0.60$ ,  $n = 15$ ,  $p < 0.001$ ) and pH value ( $r = -0.59$ ,  $n = 15$ ,  $p < 0.001$ ). The negative effect of salinity in soils of arid regions is characterized by a low amount of soil organic carbon. In other soil ecosystems under investigation no significant correlations between the parameters determined (soil pH, EC) and the number of actinomycete could be found. Davis and Williams (1970), however, reported that soil pH and moisture content could affect actinomycetes population. The lowest numbers of actinomycetes were observed at high pH values and low moisture content.



**Figure 2:** Effect of climate on actinomycetes population in forest land columns with identical letters are not statistically different ( $p < 0.05$ )

In forest soils, the number of actinomycetes was significantly dependent ( $p < 0.05$ ) on climate (Figure 2). A significantly higher population was observed under semi-arid conditions compared to all other climatic zones (humid, sub-humid, Mediterranean). Actinomycetes are not affected by semi-dry conditions as the majority of other soil bacteria and they prefer a relatively low level of moisture (Alexander, 1977). This property of actinomycetes might be due to their sporulation capability under drought conditions (El-Tarabily and Sivasithamparam, 2006). The effect of climate on actinomycetes population in the other soil ecosystems under investigation was not significant.

Inositol phosphates, phospholipids and nucleic acids are the dominant forms of organic phosphorylated compounds in soils (Quiquampoix and Mousain 2005) and inositol phosphates have been reported to be the main form of organic phosphorus (Po) in soil (Dalal, 1977). Because actinomycetes are widely distributed in different soil ecosystems, it has been assumed that they are capable of hydrolyzing phytate (myo-inositol (1,2,3,4,5,6) hexakisphosphate).

Very recently Ghorbani-Nasrabadi *et al.* (2012) gave proof for the validity of this assumption; all strains of actinomycetes under investigation were capable of producing extracellular phytate-degrading activity. Finally 67 isolates were confirmed to dephosphorylate phytate in liquid culture. Among these isolates 29, 36 and 35 % originated from irrigated/cultivated, forest and rain-fed lands, respectively. According to phytate hydrolysis during fermentation, the actinomycetes were classified into 3 different groups. Members of Group 1 were reported as poor phytate degraders, members of group 2 produced considerable amounts of InsP3 to InsP5 during fermentation in the presence of phytate and members of group 3 were very efficient in phytate degradation. Due to the limited knowledge on regulation of phytase synthesis in actinomycetes a number of studies with a limited number of isolates under laboratory conditions were performed to obtain information about extracellular factors affecting the production of phytase and other phosphatases.

**Table 1:** Effect of different nitrogen sources in the growth medium on phytate degradation with isolate No. 63 after 3 days of fermentation

N-source	InsP <sub>6</sub> [%]	InsP <sub>5</sub> [%]	InsP <sub>4</sub> [%]	InsP <sub>3</sub> [%]
(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	89.98	7.96	0.54	1.52
NaNO <sub>3</sub>	91.98	6.47	trace	1.55
NH <sub>4</sub> NO <sub>3</sub>	92.83	4.09	1.98	1.10
malt extract	93.46	4.98	0.66	1.10
meat extract	50.50	44.90	2.78	0.90
soybean meal	92.22	6.26	trace	1.12
arginine	49.48	45.76	2.72	1.94

Detection limit for allmyo-inositol phosphates was determined to be 10 µmol/L. The initial sodium phytate concentration was 4 g/L (4329 µmol/L).

### 3.2. Effect of different nitrogen and carbon sources on the phytate degradation

Because it is well established that bacterial growth and production of phytate-degrading enzyme could be dependent on the composition of the growth medium (Fredrikson *et al.*, 2002), the effect of the substitution of arginine or glycerol in the standard MGA medium on the phytate-degrading capability of the isolate with the highest extracellular phytate-degrading activity (No. 63) was investigated. Arginine was replaced by some inorganic (ammonium sulfate, ammonium

nitrate, sodium nitrate) and some organic (malt extract, soybean meal) nitrogen sources and glycerol by simple sugars such as glucose, fructose, sucrose, lactose, and maltose and complex carbon sources such as pea flour and maltodextrin. A clear dependence of phytate degradation in the fermentation broth within 3 days of fermentation on the nitrogen source (Table 1) as well as the carbon source (Table 2) was observed. In general, substitution of arginine with the organic and inorganic nitrogen sources studied resulted in a significant reduction of phytate dephosphorylation (Table 1).



**Table 2:** Effect of different nitrogen sources in the growth medium on phytate degradation with isolate No. 63 after 3 days of fermentation

C-source	InsP <sub>6</sub> [%]	InsP <sub>5</sub> [%]	InsP <sub>4</sub> [%]	InsP <sub>3</sub> [%]
glucose	89.76	9.22	0.32	0.71
maltose	91.96	6.48	traces	1.56
lactose	87.05	11.15	traces	1.80
fructose	81.11	16.21	1.25	1.43
sucrose	92.70	5.96	traces	1.34
maltodextrine	72.40	24.76	1.65	1.19
pea flour	60.28	32.96	4.58	1.73
glycerol	49.58	45.76	2.72	1.94

detection limit for allmyo-inositol phosphates are indicated in  $\mu\text{mol/l}$  was determined to be  $10 \mu\text{mol/L}$ . The initial sodium phytate concentration was  $4 \text{ g/L}$  ( $4329 \mu\text{mol/L}$ ).

With ammonium sulfate and sodium nitrate about 60% and with ammonium nitrate, malt extract and soybean meal about 40% of phytate hydrolysis compared to arginine was determined. This observation is in accordance with Ramachandran *et al.* (2005). They reported a reduction in fungal phytase production in the presence of sodium nitrate and ammonium sulfate. Substitution of arginine with meat extract however, did not have any effect on the capability of isolate No.63 to degrade the phytate present in the growth medium

(Table 1). Replacing glycerol in the standard MGA medium by glucose, maltose and sucrose did almost completely suppress phytate dephosphorylation (Table 2). With fructose about 70% and with lactose and maltodextrin about 50% of phytate hydrolysis compared to glycerol was observed. Substitution of glycerol with pea flour however, did not have any effect on the capability of isolate No. 63 to degrade the phytate present in the growth medium.

**Table 3:** Acid and alkaline phosphatase activity in different actinomycetes isolates in standard MGA medium

Isolate No.	Acid phosphatase (mU ml <sup>-1</sup> )	Alkaline phosphatase (mUml <sup>-1</sup> )
39	13.5 a	18.16 b
43	1.12 f	0.59 e
44	0.92 g	1.37 d
45	5.23 d	1.37 d
46	1.56 e	1.32 d
47	0.99fg	0
50	8.27 b	27.59 a
63	6.74 c	3.02 c

numbers with identical letters are not significantly ( $p < 0.05$ ) different

### 3.3. Acid and alkaline phosphatase activity

In order to make the organic phosphorus pool in soil available for plants and soil microorganism, the organically bound phosphate must be released. Because plant roots do not exude significant amounts of phosphatases including phytases, it was assumed that soil microorganisms may play an important role in this process. Release of organically bound phosphate is a concerted action of phytases and other phosphatases and therefore the eight actinomycetes isolates, among the 67 isolates confirmed by Ghorbani-Nasrabadi *et al.* (2012) to dephosphorylate phytate in liquid culture, with the highest capability to produce

extracellular phytases were studied in respect to their extracellular acid and alkaline phosphatase activity (Table 3). A huge variation in the capacity to produce acid and alkaline phosphatases among the different isolates was observed. Neither the capacity to produce acid nor the capacity to produce alkaline phosphatases did correlate with the phytate-degrading capacity. The highest extracellular phytate-degrading activity was observed with isolate No. 63, whereas the highest acid phosphatase activity was found with isolate No. 39 and the highest alkaline phosphatase activity with isolate No. 50.

**Table 4:** Effect of media composition on acid and alkaline phosphatase activity in isolate No. 63

Medium	Acid phosphatase (mUml <sup>-1</sup> )	Alkaline phosphatase (mUml <sup>-1</sup> )
SPG	34.0 a	5.01 a
MGA	6.74 b	3.02 b
ISP2	0	0
LB+ rice bran	3.34 c	0.75 c
Corn starch	1.97 d	0

numbers with identical letters are not significantly ( $p < 0.05$ ) different

#### 3.4. Effect of media composition on phosphatase activity

It is well documented that expression of phosphatases in microorganisms is strongly affected by medium composition (Pedregosa *et al.*, 1991). Different media were used to evaluate the effect of medium composition on extracellular acid and alkaline phosphatase activities of isolate No. 63 and it was confirmed that these enzyme activities are dependent on the medium used for fermentation (Table 4). However, acid and alkaline phosphatase activities were affected differently by medium composition. In SPG, for example, acid phosphatase activity was 5-fold higher compared to ISP2, whereas alkaline phosphatase activity was only 1.7-fold higher. In MGA both acid and alkaline phosphatase activities were below the detection limit of the enzyme activity assays.

#### 4. Conclusion

Actinomycetes are one of the predominant members of soil microbial communities and they have beneficial roles in soil nutrients cycling and agricultural productivity (Elliot and Lynch, 1995). With the exception of pasture, soil factors such as pH and salinity do not significantly affect actinomycetes population in the different soil ecosystems studied. The effect of salinity on microbial biomass and activity is not uniform and the observed differences may be due to the composition of the pool of soluble ions as well as the presence or absence of plant and agricultural activities. In this study no significant difference in the number of actinomycetes was observed comparing irrigated and rain-fed land uses, even though electrical conductivity was significantly different ( $p < 0.05$ ). This observation supports the conclusion of Chowdhury *et al.* (2011) that the combined effects of matric and osmotic potential in saline soils is more important for microbial biomass

than electrical conductivity. In addition, the knowledge on the participation of actinomycetes in hydrolysis of organic phosphorylated compounds is extremely limited. It went clear that production of phytases and other phosphatases is strongly dependent on the actinomycetes strain as well as media composition. Furthermore, production of phytases, acid phosphatases and alkaline phosphatases was shown to be regulated independently. However, the importance of actinomycetes in dephosphorylation of soil organic compounds needs to be elucidated in further studies.

## References

- Alexander, M. 1977. Introduction to soil microbiology, 2nd edition, Krieger Publishing Company, USA.
- Bossio, D.A., Girvan, M.S., Verchot, L., Bullimore, J., Borelli, T., Albrecht, A., Scow, K.M., Ball, A.S., Pretty, J.N., Osborn, A.M. 2005. Soil microbial community response to land use change in an agricultural landscape of Western Kenya. *Microbial Ecology*. 49, 50–62.
- Burck, I.C., Yonker, C.M., Parton, W.J., Cole, C.V., Flach, K., Schimel, D.S. 1989. Texture, climate, and cultivation effects on soil organic matter content in USA grassland soils. *Journal of Soil Science Society of America*. 53, 800–805.
- Chowdhury, N., Marschner, P., Burns, R.G. 2011. Soil microbial activity and community composition: impact of changes in matric and osmotic potential. *Soil Biology and Biochemistry*. 43, 1229-1236.
- Dalal, R.C. 1977. Soil organic phosphorus. *Advances in Agronomy*. 29, 83–117.
- Davies, F.L., Williams, S.T. 1970. Studies on the ecology of actinomycetes in soil I. The occurrence and distribution of actinomycetes in a pine forest soil. *Soil Biology and Biochemistry*. 2, 227-238.
- Elliot, L.F., Lynch, J.M. 1995. The international workshop on establishment of microbial inocula in soils: cooperative research project on biological resource management of the Organization for Economic Cooperation and Development (OECD). *American Journal of Alternative Agriculture*. 10, 50-73.
- El-Nakeeb, M.A., Leachevalier, H.A. 1963. Selective isolation of aerobic actinomycetes. *Applied Microbiology*. 11, 75-77.
- El-Tarabily, K.A., Sivasithamparam, K. 2006. Non-streptomycete actinomycetes as biocontrol agents of soil-borne fungal plant pathogens and as plant growth promoters. *Soil Biology and Biochemistry*. 38, 1505–1520.
- Fierer, N., Carney, K.M., Horner-Devine, M.C., Megonigal, J.P. 2009. The biogeography of ammonia-oxidizing bacterial communities in soil. *Microbial Ecology*. 58, 435-445.
- Fredrikson, M., Andlid, T., Haikara, A., Sandberg, A.S. 2002. Phytate degradation by micro-organisms in synthetic media and pea flour. *Journal of Applied Microbiology*. 93, 197-204.
- Gee, G.W., Bauder, J.W. 1986. Particle-size analysis, in: A. Klute (Ed.), *Methods of Soil Analysis, Part 1. Physical and Mineralogical Methods*, 2nd edition.
- Ghorbani-Nasrabadi, R., Greiner, R., Alikhani, H. A., Hamed, J. 2012. Identification and determination of extracellular phytate-degrading activity in actinomycetes. *World Journal of Microbiology and Biotechnology*. 28, 2601-2608.
- Goodfellow, M., Williams, S.T. 1983. Ecology of actinomycetes. *Annual Review of Microbiology*. 37, 189-216.
- Greiner, R., Jany, K.D. 2003. Purification and characterisation of homogeneous acid phosphatase without phytate-degrading activity from non-germinated buck wheat (*Fagopyrum esculentum*) seeds. *Journal of Food Biochemistry*. 27, 197-220.

- Greiner, R., Konietzny, U., Jany, K.D. 2007. Purification and properties of a phytase from rye. *Journal of Food Biochemistry*. 22, 143-161.
- Hamdali, H., Hafidi, M., Virolle, M.J., Ouhdouch, Y. 2008. Growth promotion and protection against damping-off of wheat by two rock phosphate solubilizing actinomycetes in a P-deficient soil under greenhouse conditions. *Applied Soil Ecology*. 40, 510-517.
- Harrison, A.F. 1987. Soil organic phosphorus. A review of world literature. Oxford CAB International.
- Khormali, F., Ajami, M., Ayoubi, S., Srinivasarao, Ch., Wani S.P. 2009. Role of deforestation and hillslope position on soil quality attributes of loess-derived soils in Golestan province, Iran. *Agriculture, Ecosystems & Environment*. 134, 178-189.
- Lauber, C.L., Hamady, M., Knight, R., Fierer, N. 2009. Pyrosequencing-based assessment of soil pH as a predictor of soil bacterial community structure at the continental scale. *Applied and Environmental Microbiology*. 75, 5111-5120.
- McCarthy, A.J., Williams, S.T. 1992. Actinomycetes as agents of biodegradation in the environment - a review. *Gene*. 115, 189-192.
- McLean, E.O. 1982. Soil pH and lime requirement, in: A.L. Page, A.L., R.H. Miller, D.R. Keeney (Eds.), *Methods of Soil Analysis, Part 2. Chemical and Microbiological Properties*, 2nd edition.
- Nardi, S., Cocheri, G., Dell'Agnoia, G. 1996. Biological activity of humus, in: A. Piccolo (Ed.), *Humic Substances in Terrestrial Ecosystems*. Elsevier, Amsterdam pp. 361-406.
- Nelson, D.W., Sommers, L.E. 1982. Total carbon, organic carbon, and organic matter, in: A.L. Page, R.H. Miller, D.R. Keeney (Eds.), *Methods of Soil Analysis, Part 2. Chemical and Microbiological Properties*. 2nd edition.
- Ouahmane, L., Thioulouse, J., Hafidi, M., Prin, Y., Ducouso, M., Galiana, A., Plenchette, C., Kisa, M., Duponnois, R. 2007. Soil functional diversity and P solubilization from rock phosphate after inoculation with native or mycorrhizal fungi. *Forest Ecology and Management*. 241, 200-208.
- Pedregosa, A.M., Pinto, F., Monistrol, I.F., Laborda, F. 1991. Regulation of acid and alkaline phosphatases of *Cladosporium cucumerinum* by inorganic phosphate. *Mycological Research*. 95, 720-724.
- Peix, A., Rivas-Boyer, A.A., Mateos, P.F., Rodriguez-Barrueco, C., Martinez-Molina, E., Velazquez, E. 2001. Growth promotion of chickpea and barley by a phosphate solubilizing strain of *Mesorhizobium mediterraneum* under growth chamber conditions. *Soil Biology and Biochemistry*. 33, 103-110.
- Puget, P., Lal, R. 2005. Soil organic carbon and nitrogen in a Mollisol in central Ohio as affected by tillage and land use. *Soil and Tillage Research*. 80, 201-213.
- Quiquampoix, H., Mousain, D. 2005. Enzymatic hydrolysis of organic phosphorus, in: B.L. Turner, E. Frossard, D.S. Baldwin (Eds), *Organic phosphorus in the environment*, CABI, UK. pp. 89-112.
- Ramachandran, S., Krishnan, R., Nampoothiri, K.M., Szackacs, G., Pandey, A. 2005. Mixed substrate fermentation for the production of phytase by *Rhizopus* spp. using oil cakes as substrates. *Process Biochemistry*. 40, 1749-1754.
- Rodriguez, H., Fraga, R. 1999. Phosphate solubilizing bacteria and their role in plant growth promotion. *Biotechnology Advances*. 17, 319-339.
- Ryan, J., Garabet, S., Rashid, A., El Gharous, M. 1999. Assessment of soil and plant analysis laboratories in the West Asia-North Africa Region. *Communications in Soil Science and Plant Analysis*. 30, 885-894.

- Sandberg, A.S., Ahderinne, R. 1986. HPLC method for determination of inositol tri-, tetra-, penta-, hexaphosphates in foods and intestinal contents. *Journal of Food Science*. 51, 547-550.
- SAS Institute.2000. SAS Online Doc. Version 8. SAS Inst., Cary, NC. available at: <http://v8doc.sas.com/sashtml/> (accesses 02.04.08; verified 23.02.04).
- Van Vuuren, D.P., Bouwman, A.F., Beusen, A.H.W. 2010. Phosphorus Demand for 1970–2100 Period: A Scenario Analysis of Resource Depletion, *Global Environmental Change*. 20, 428-439.
- Williams, S.T., Wellington, E.M.H. 1982. Principles and problems of selective isolation of microbes, in: J.D. Bullock, L.J. Nisbet, D.J. Winstanley (Eds.), *Bioactive Microbial Products 1: Search and Discovery*, Academic Press, UK pp. 9-26.
- Wolf, D.C., Wagner, G.H. 2005. Carbon transformations and soil organic matter formation, in: D.M. Sylvia, J.J. Fuhrmann, P.G. Hartel, D.A. Zuberer (Eds.), *Principles and applications of soil microbiology*. Pearson Prentice Hall, USA pp. 285-332.
- Zheng, Z., Zeng, W., Huang, Y., Yang, Z., Li, J., Cai, H., Su, W. 2000. Detection of antitumor and antimicrobial activities in marine organism associated actinomycetes isolated from the Taiwan Strait, China. *FEMS Microbiology Letters*. 188, 87-91.